SYNTHESES AND ANTITUMOR ACTIVITIES OF 7-O-(6-DEOXY-2-O-METHYL-α-L-TALOPYRANOSYL)-DAUNOMYCINONE AND -ADRIAMYCINONE

Sir:

As described in preceding papers¹⁻⁴⁾, we prepared several fluorine-containing daunorubicin and doxorubicin analogs and showed that a combination of α -side 2'-fluoro and α -side 3'-hydroxyl substituents in the sugar moiety gave the most promising compounds in terms of anticancer activity. In this paper

we describe the synthesis and activities of analogs 10 and 14 having an α -side 2'-methoxyl group with α -L-talopyranoside structure instead of the 2'-fluorine. The methoxyl group is electron-attractive similar to the fluorine atom although the action is weaker than the fluorine atom, therefore, 10 and 14 should show similar activities to those of the corresponding fluorine analogs. Interestingly steffimycins⁵⁾ and nogalamycin⁶⁾, the anthracycline antibiotics of natural origin, have an α -side 2-methoxyl group at C-2 of the sugar moieties.

3,4-Di-O-acetyl-6-deoxy-2-O-methyl- α -L-talopyranosyl bromide (7) has been prepared

17 $R_1 = SO_3Na$

 $R_2 = H$

from methyl 3,4-O-isopropylidene- β -L-fucopyranoside (1)2) via 6 steps. Oxidation of 1 with pyridinium chlorochromate (in CH2Cl2 in the presence of powdered Molecular Sieves 3A) gave the 2-oxo derivative (2, 73%), mp $64 \sim 65.5$ °C, which, on reduction with LiAlH₄ (in THF, 0°C to room temperature), gave the β -L-talopyranoside (3, 83%): MP $91 \sim 92^{\circ}$ C; $[\alpha]_{D}^{21} + 17^{\circ}$ (CHCl₃); ¹H NMR (CDCl₃) δ 4.36 (1H, d, 1-H), 3.76 (1H, ddd, 2-H); $J_{1,2}=1.5$ Hz, $J_{2,3}=5$ Hz, $J_{2,OH}$ =8.5 Hz. Methylation of 3 with MeI (in MeCN in the presence of Ag₂O, 80°C) gave the 2-O-methyl derivative (4, 89%): MP 116 \sim 118°C; ¹H NMR (CDCl₃) δ 3.58 and 3.53 (each 3H, s, OCH₃). Deacetonation (aq 80% AcOH, 80°C) gave 5 (98%): MP $102.5 \sim 104$ °C (needles); $[\alpha]_D^{23}$ +91° (CHCl₃). Acetylation of 5 $(Ac_2O - H_2SO_4, 1:0.03 \text{ in } MeNO_2, 0^{\circ}C)$ gave the 1,3,4-tri-O-acetyl-6-deoxy-2-O-methyl- α -Ltalopyranose (6, 64%): MP 128.5~130.5°C; $[\alpha]_D^{23}$ -75° (CHCl₃); ¹H NMR (CDCl₃) δ 6.27 (1H, d, 1-H), 3.53 (3H, s, OCH₃), 2.19, 2.13 and 2.10 (each 3H, s, Ac). Bromination of 6 (TiBr₄ in CH₂Cl₂-EtOAc, 10:1, 27°C) gave the 1bromide (7, syrup, 90%).

Coupling of 7 with daunomycinone was carried out by a Koenigs-Knorr type of reaction (HgO(yellow) - HgBr₂ - powdered Molecular Sieves 3A, in CH₂Cl₂) to give the desired α -L-glycoside (8, 69%), a red solid: $[\alpha]_D^{20} + 202^{\circ}$ (c, 0.1, CHCl₃), along with its β -L-anomer (9, 9%): $[\alpha]_D^{23} + 368^{\circ}$ (c 0.02, CHCl₃). Deprotection of 8 (0.2 N aq NaOH, 0°C to room temperature) gave 7-O-(6-deoxy-2-O-methyl- α -L-talopyranosyl)daunomycinone (10, 73%), a red

solid: $[\alpha]_{0}^{22} +132^{\circ}$ (c 0.05, CHCl₃ - MeOH, 1:1); ¹H NMR (CDCl₃) δ 5.59 (1H, br s, 1'-H), 3.52 (3H, s, 2'-OCH₃), 3.41 (1H, dt, 2'-H), 2.41 (3H, s, 14-CH₃): $J_{1',2'} = J_{2',4'} = \sim 1.5 \text{ Hz}$, $J_{2',3'} = 3.5$ Hz. The α -L-configuration of 10 was confirmed by the observation of nuclear Overhauser effect (NOE) between 1'-H and 2'-OCH3 in its 1H-1H NOE difference spectrum. The corresponding 14-hydroxy compound (14) was obtained by coupling of 7 with 14-O-tert-butyldimethylsilyladriamycinone (11)7) in a similar manner as described for 8. The coupled compound 12 (52%), a red solid; $[\alpha]_D^{24} + 166^\circ$ (c 0.05, CHCl₃) was deacetylated (with MeONa - MeOH, to give 13 (98%)) and desilylated (aq 80% AcOH, 80°C) to give 14 (72%), a red powder: $[\alpha]_{D}^{21}$ +129° (c 0.02, CHCl₃ - MeOH, 1:1); ¹H NMR $(CDCl_3)$ δ 5.60 (1H, d, 1'-H), 4.75 (2H, d, CH₂OH), 3.52 (3H, s, 2'-OCH₃).

In order to obtain a water-soluble derivative of 14, which is sparingly soluble in water, sodium salt of 14-hemisulfate (17) was prepared. A reaction intermediate (12) was desilylated (ag 80% AcOH, 80°C) to give the 14-hydroxy derivative (15, 90%), $[\alpha]_D^{20}$ +190° (c 0.02, CHCl₃). Sulfation of 15 (C₅H₅N·SO₃ in DMF, 80°C, then neutralized with aq NaOH to give 16 (66%)) followed by deacetylation (MeONa -MeOH) gave the water-soluble 17 (81%): $[\alpha]_D^{21}$ +208° (c 0.01, CHCl₃ - MeOH, 1:1); $[\alpha]_{\rm D}^{19}$ -82° (c 0.02, H₂O); ¹³C NMR (DMSO-d₆) δ 208.7 (C-13), 67.2 (C-14) (cf. 14: δ 213.7 (C-13), 63.7 (C-14); Anal Calcd for C28H29NaO16S. 2H₂O: C 47.19, H 4.67, S 4.50. Found: C 47.47, H 4.86, S 4.60.

Table 1. Antitumor activities (T/C, %; 60 days survivor numbers/treated numbers of mice) of 10 and 14 in comparison with daunorubicin (DNR) and doxorubicin (DOX) on L1210.

Compound		Dose (µg/mouse/day)										
		400	200	100	50	25	12.5	6.25	3.13	1.56	0.78	0.39
10	T/C (%)	106*	>463*	> 389	137	109	94		-			
	Survivor	0/4	2/4	1/4	0/4	0/4	0/4					
14	T/C(%)		101*	153*	>477	>494	172	230	141	107	95	103
	Survivor		0/4	0/4	2/4	2/4	0/4	0/4	0/4	0/4	0/4	0/4
DNR · HCl	T/C(%)			117*	151*	193	166	133	130		,	,
	Survivor			0/4	0/4	0/4	0/4	0/4	0/4			
DOX-HCl	T/C(%)			177*	273*	330	208	132	140			
	Survivor			0/4	0/4	0/4	0/4	0/4	0/4			

Leukemia L1210 cells (10^5) were inoculated into CDF₁ mice ($20\pm1\,\mathrm{g}$) intraperitoneally. Drugs were administered daily, starting 24 hours after inoculation, from day-1 to -9, intraperitoneally. * Toxic.

As shown in Table 1, 10 and 14 showed much improved antitumor activities in comparison with daunorubicin and doxorubicin. The sodium salt of 14-hemisulfate (17) was, however, inactive at doses up to 200 µg/mouse/day. These results indicate that introduction of a methoxyl group at C-2′, which strengthens a glycosyl bond as the fluorine atom does, is also expected to give compounds with similarly promising biological activities as those of the corresponding 2′-fluorine compounds.

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